



# Faculté Polytechnique

## 2ND CAPARICA CHRISTMAS CONFERENCE ON

# SAMPLE TREATMENT

5TH - 7TH OF DECEMBER 2016 CAPARICA, PORTUGAL

## Morphological and microstructural observation of Duplex electroless coatings by TEM after sample selection by FIB.

V. Vitry, L. Bonin, L. Mallet



**materials**  
UMONS RESEARCH INSTITUTE  
FOR MATERIALS SCIENCE  
AND ENGINEERING



Dr Ir Vitry Véronique

Veronique.vitry@umons.ac.be



# Electroless nickel coatings?

Reduction of nickel salts in aqueous solution, no external current

Constant thickness

All kinds of substrates

1946 (KaNiGen)

Deposition of metal (reduction)



Oxydation of a metal



External current source



Oxydation of chemical agent



Thermodynamic condition



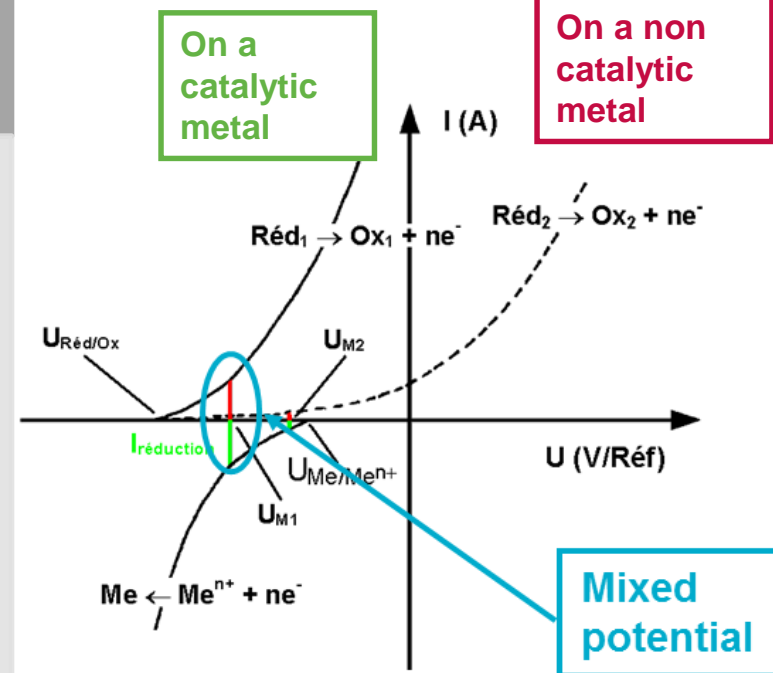
- Spontaneous reaction possible only if

$$U_{\text{Ni/Ni}^{2+}} - U_{\text{Red/Ox}} > 0$$

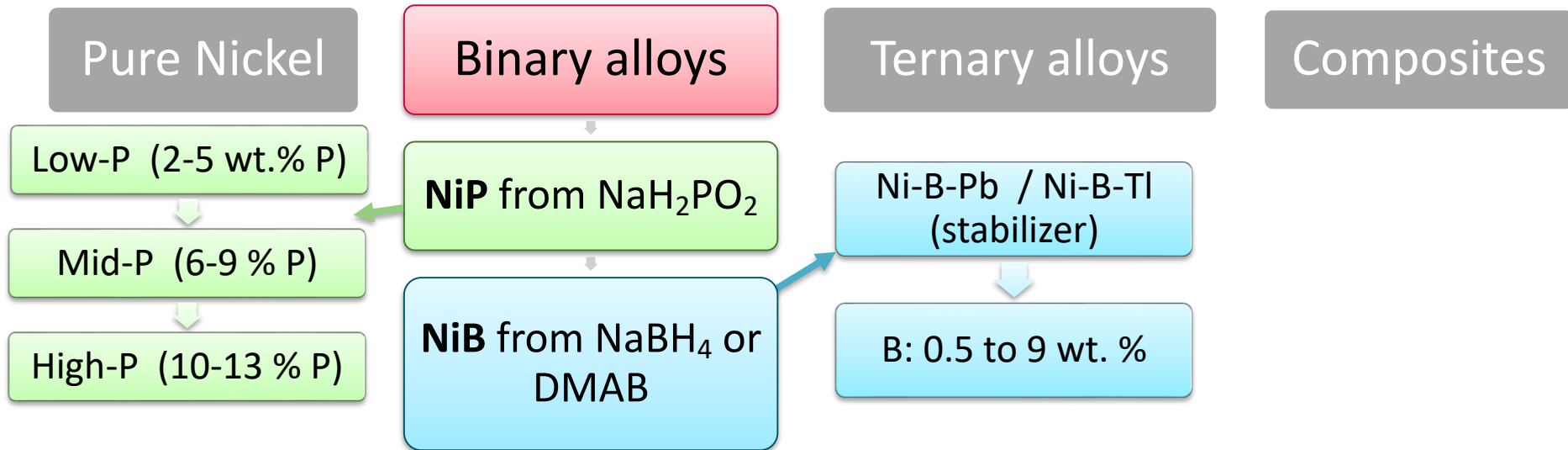
$$\Rightarrow U_{\text{Ni/Ni}^{2+}} > U_{\text{Red/Ox}}$$

Kinetic condition

- Deposition occurs at mixed potential
- Catalytic activity: Plating rate influenced by over-potential
  - On the substrate
  - On the pre-deposited coating



# Electroless nickel coatings?



## Applications



	Ni-P (2-13 wt.%)	Ni-B (1-8 wt.%)
Hardness	500-700 $\text{hv}_{100}$	up to 900 $\text{hv}_{100}$
Abrasion resistance	++ (TWI: 24)	+++ (TWI: 11)
Adhesion (scratch tests)	+++	+++
Corrosion resistance	+++	+
	+: fair	++: good
		+++: excellent

# Electroless nickel coatings?

Component	Role	Ni-B bath*	Ni-P bath
Metallic ions	Source of metal to be deposited	$\text{NiCl}_2 \cdot 6\text{H}_2\text{O}$	Niklad ELV 808 A/B (Mc Dermid)
Reducing agent	Source of electrons	$\text{NaBH}_4$	
Complexing agent	Complexes the metallic ions to <ul style="list-style-type: none"> <li>• increase the metallic ions solubility</li> <li>• increase the stability by avoiding precipitation</li> </ul> BUT decreases the deposition current	$\text{NH}_2\text{CH}_2$ $\text{CH}_2\text{NH}_2$	
Stabilizer	Regulate deposition speed by blocking a part of the catalytic sites	$\text{PbWO}_4$	
pH adjuster	Regulates pH	$\text{NaOH}$	

\*F. DELAUNOIS, J.-P. PETITJEAN, P. LIENARD, M. JACOB-DULIERE, Autocatalytic electroless nickel-boron plating on light alloys, Surface and Coatings Technology 124 (2000), 201-209

# Research question

NiP better for corrosion

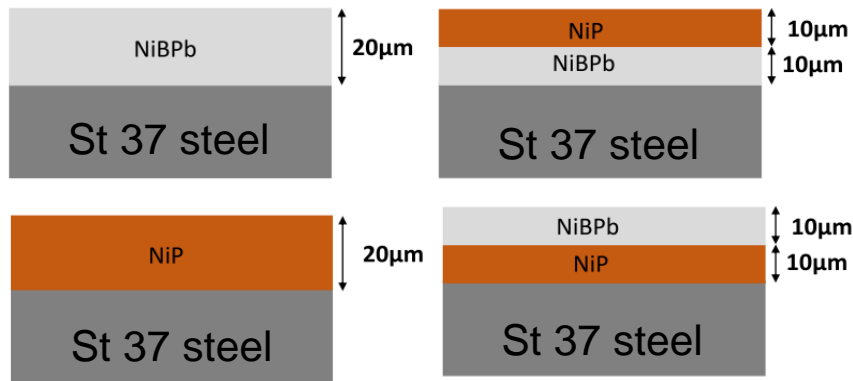


NiB better for wear and hardness

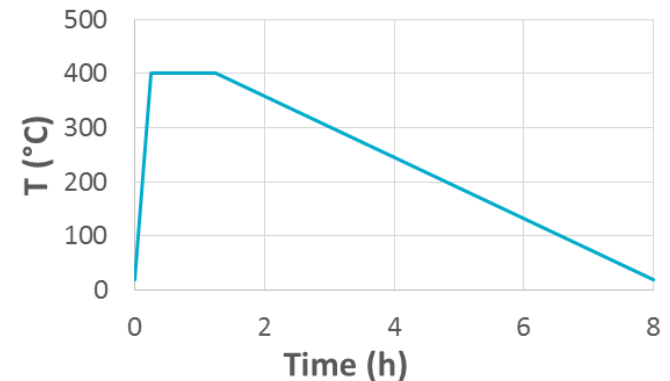


Combined (Duplex) coatings for optimised properties

## Electroless nickel deposition



## Heat treatment



Properties of Duplex coatings	As plated				Heat treated			
	NiB	NiP	NiP/NiB	NiB/NiP	NiB	NiP	NiP/NiB	NiB/NiP
Hardness	1	4	2	3	4	1	3	2
Abrasion	3	4	1	2	1	4	2	3
Wear	1	4	2	3	2	1	4	3
Scratch resistance	1	3	2	4	1	3	2	4

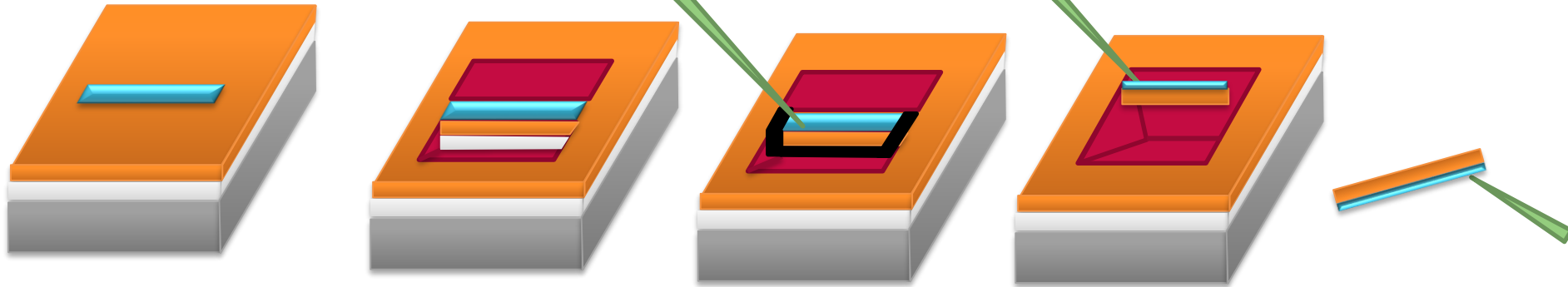
?

Influence of structure/interfaces on properties



# Experimental

'Classic' Sample preparation for TEM observation by FIB lift-out method



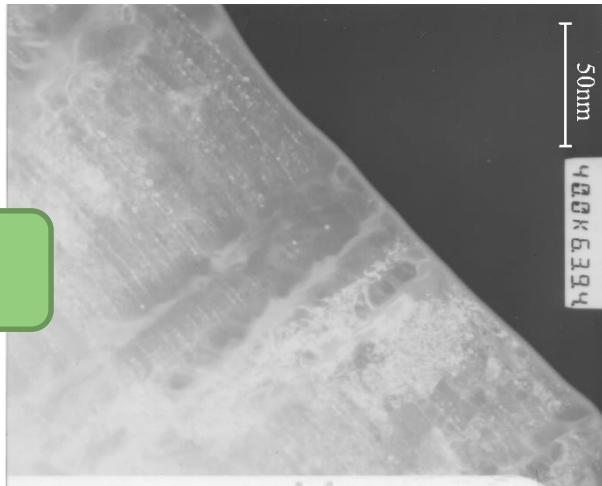
Observation of the sample by SEM and deposition of protective layer

FIB milling

Lift out

Thinning by ion milling

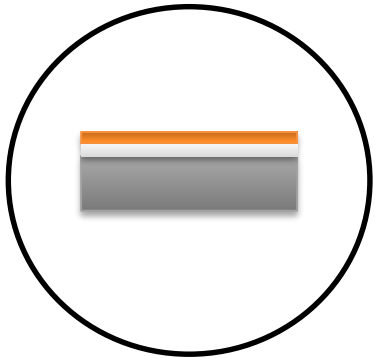
Results



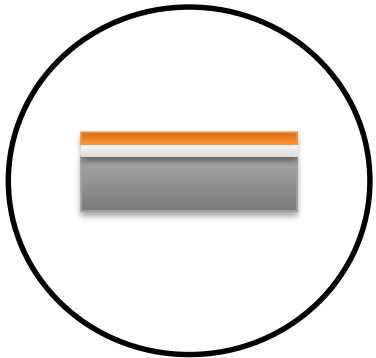
- No access to interface
- Information only about top layer

# Experimental

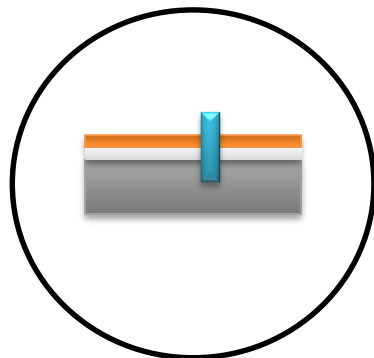
Alternate method, based on area selection lift out method



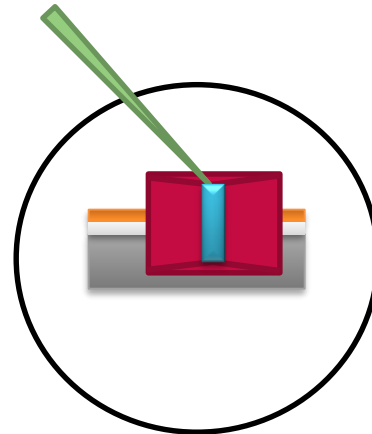
Initial sample: embedded and polished cross section



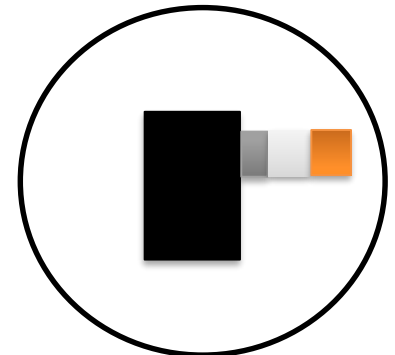
Detection of zone of interest by SEM



Deposition of protective layer on zone



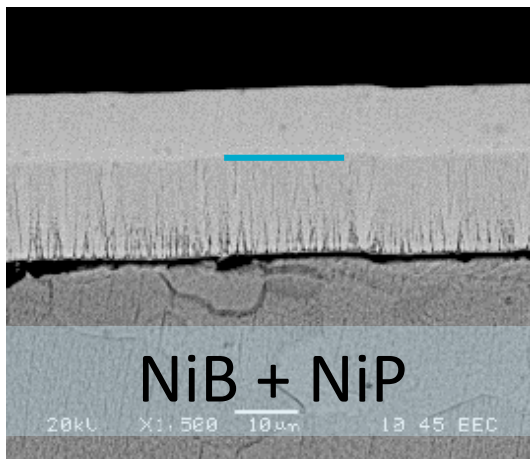
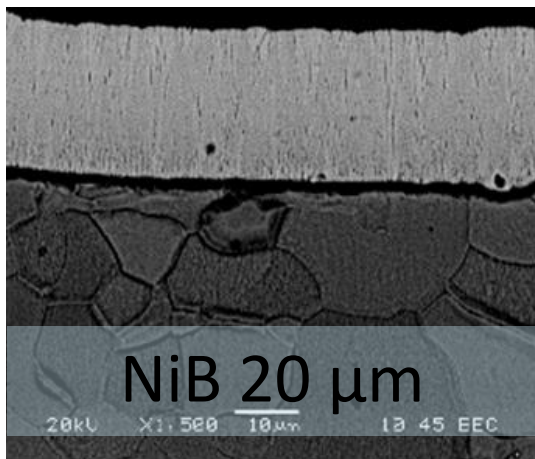
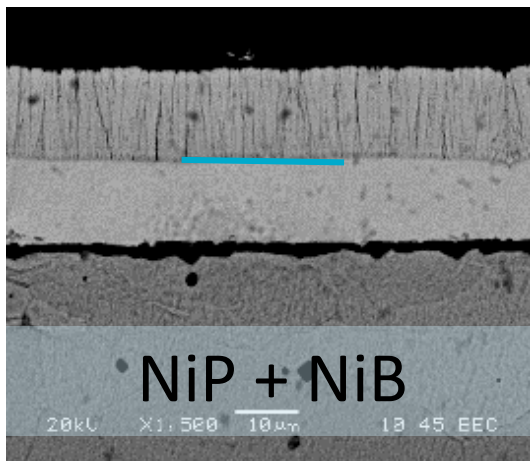
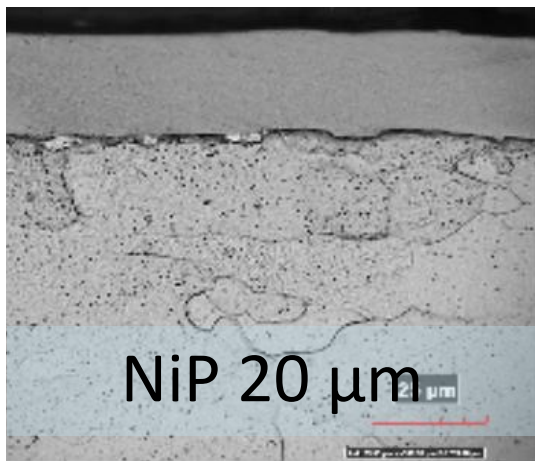
FIB milling and lift out



Thinning by ion milling

# Preliminary sample characterization

As plated samples



## Chemistry (atomic %), by GDOES:

- NiB: 6.5 – 7.5% B + 1-1.5% Pb
- NiP: 7.5 - 9.5 % P
- Balance Ni

	Ni	P	B	Pb
Ni-B/Ni-P	91,39	9,14	6,72	1,21
Ni-P/Ni-B	90,01	7,79	7,18	1,47
Ni-B	91,99		6,66	1,11
Ni-P	92,06	7,78		

## Morphology

- Clear delimitation of the various layers
- ENB columnar – ENP featureless
- No apparent influence of interface on growth.

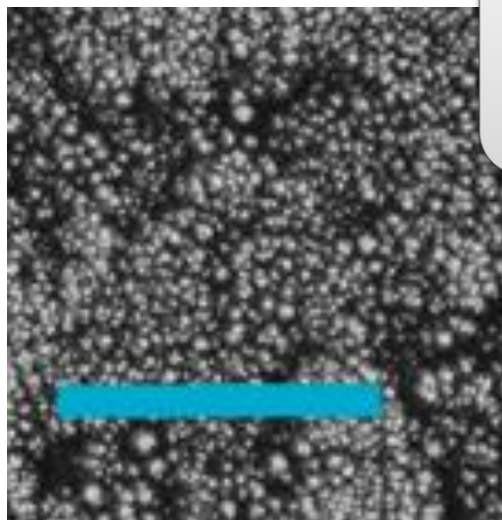


# Preliminary sample characterization

As plated samples



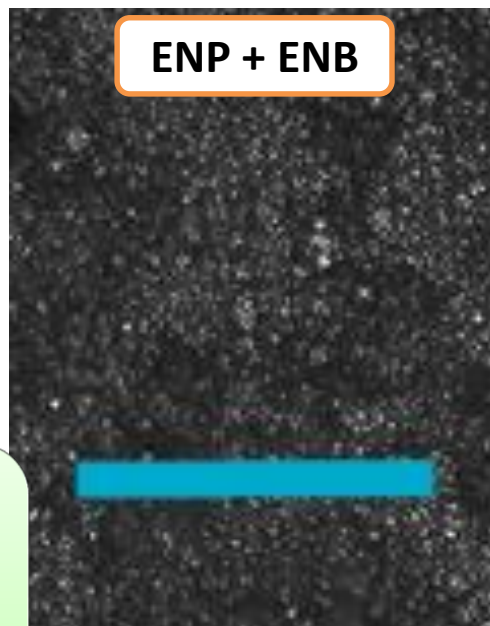
**NiP**  
(planar)



**NiB**  
(cauliflower-like)

100μm

**Duplex**  
(mixed features)



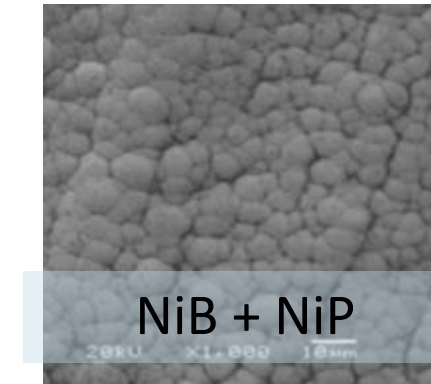
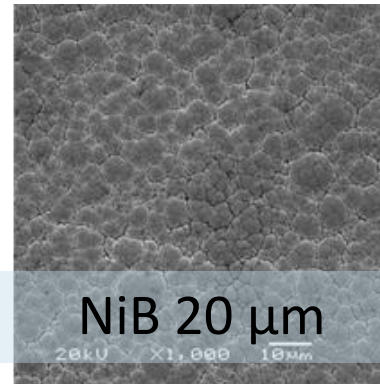
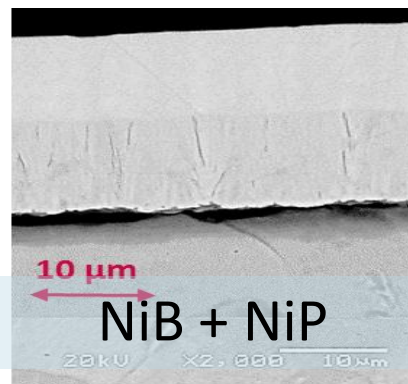
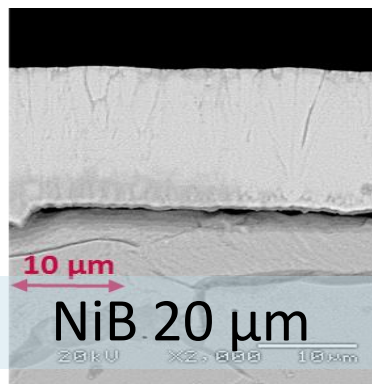
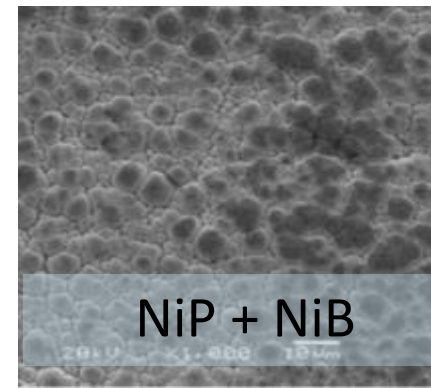
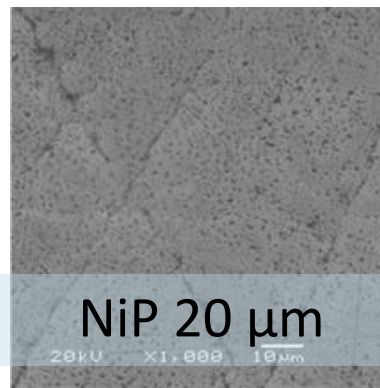
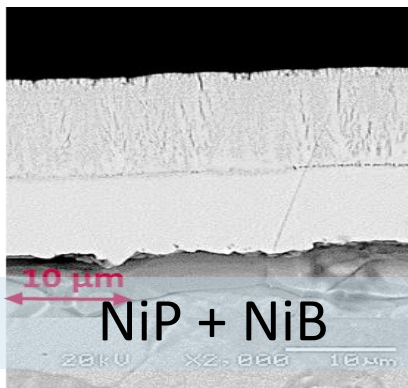
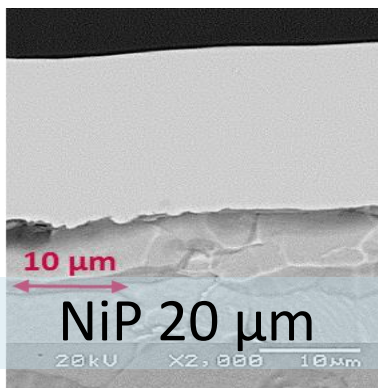
ENP + ENB



ENB + ENP

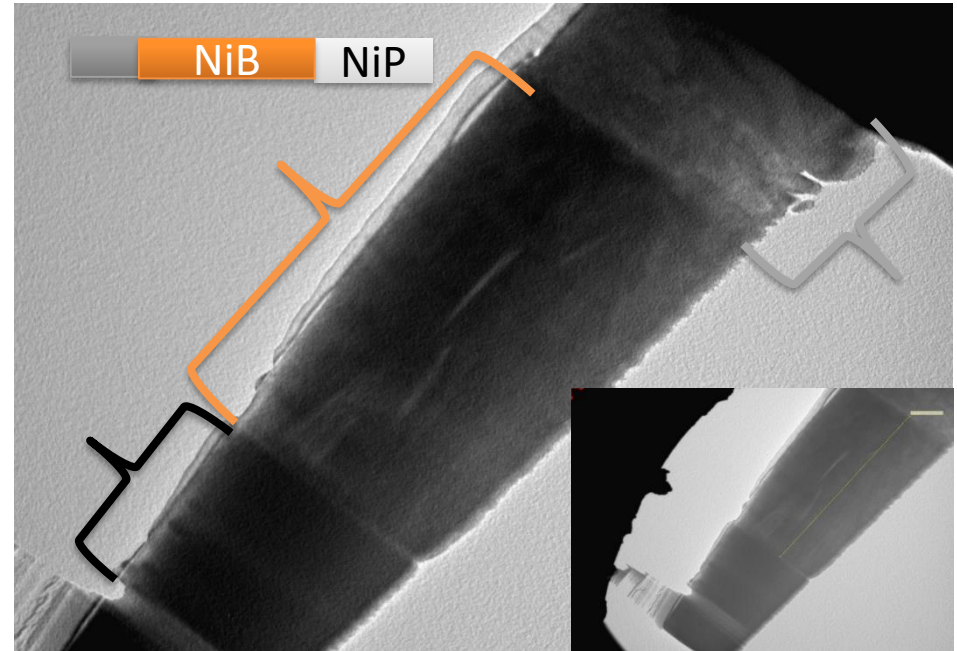
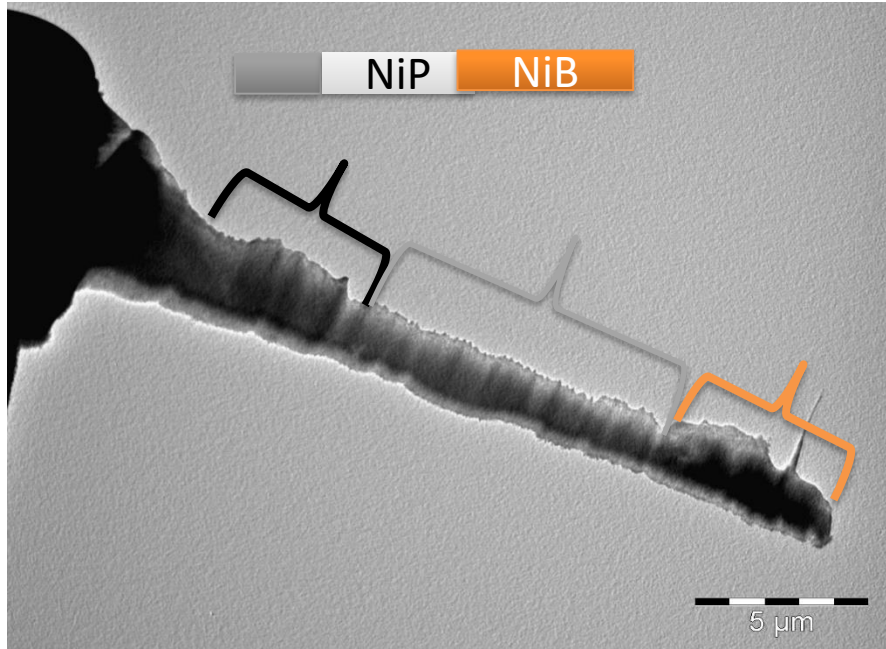
# Preliminary sample characterization

## Heat treated samples



No significant morphological modification

# SEM observation of 'lift out samples'

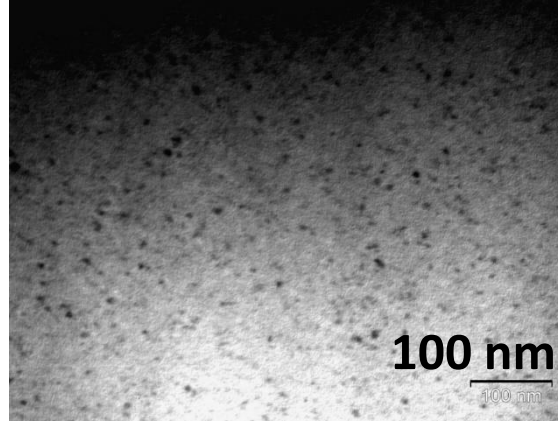
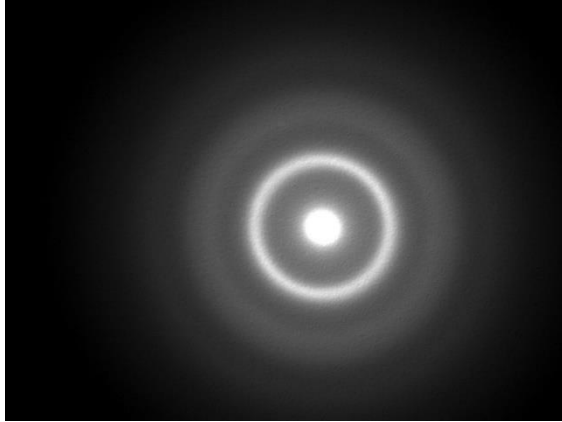


Clear view of all interfaces



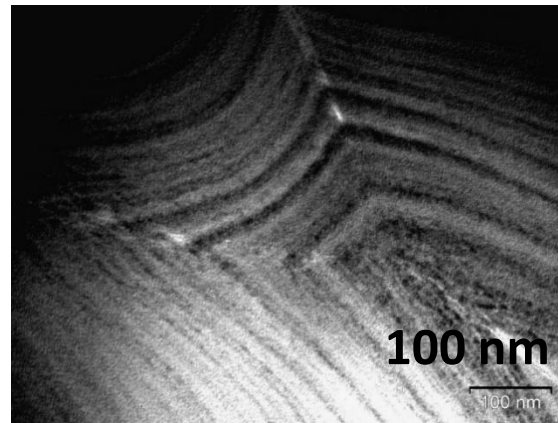
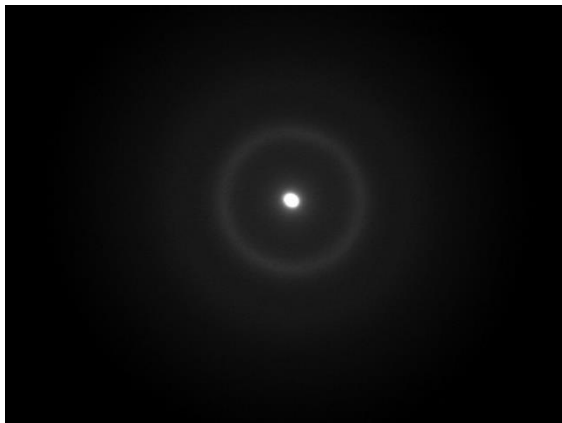
# TEM observation

## 'Bulk' of NiP layers



Nanocrystalline  
supersaturated  
nickel

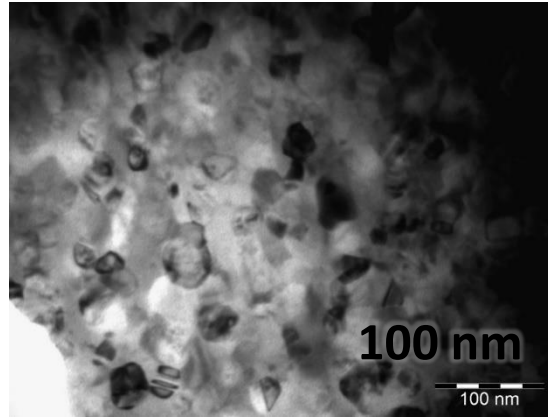
## 'Bulk' of NiB layers



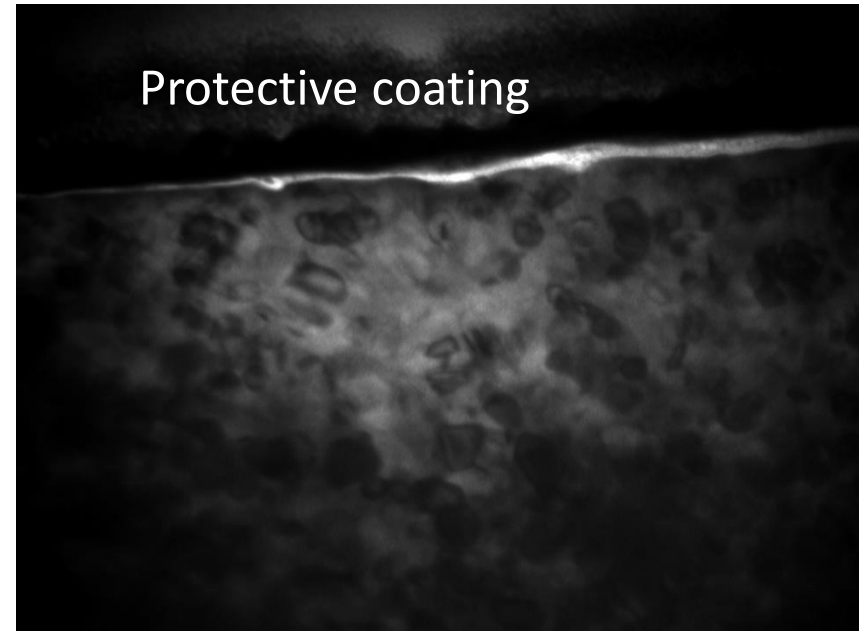
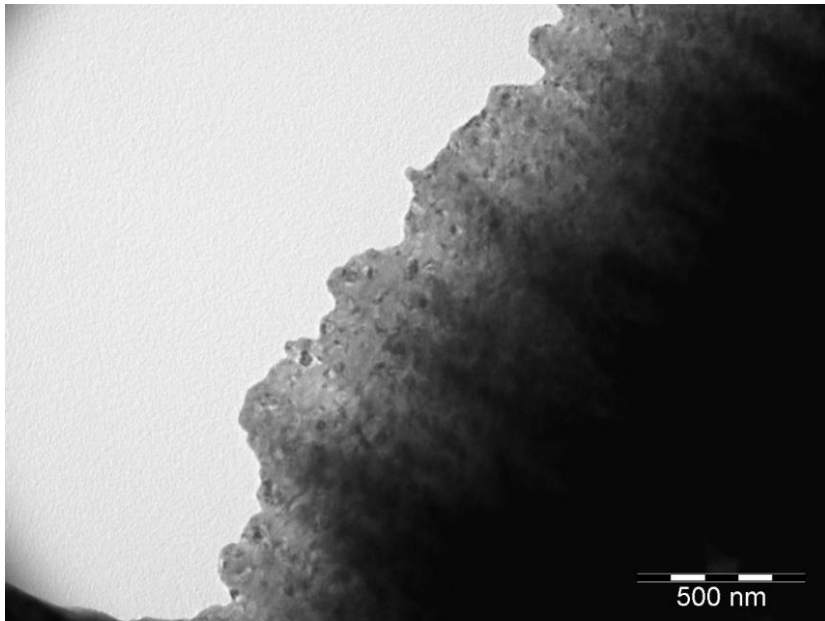
Quasi amorphous  
supersaturated  
nickel  
Layered growth

# TEM observation

'Bulk' of NiP layers – heat treated



Bigger nanocrystalline  
grains  
 $\text{Ni}_3\text{P}$  phase

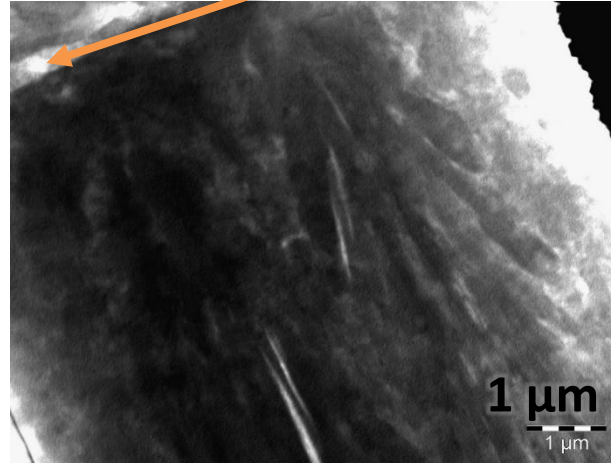
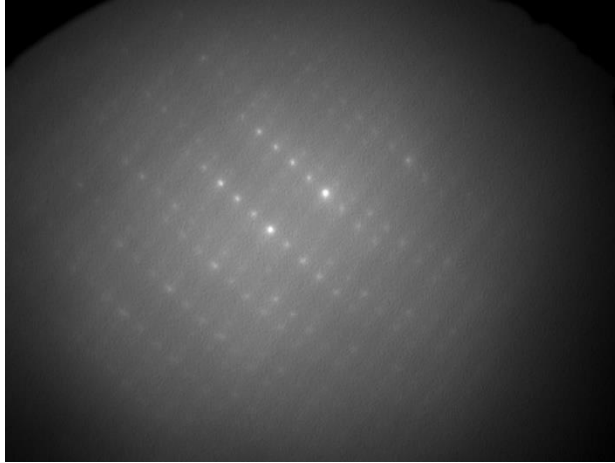




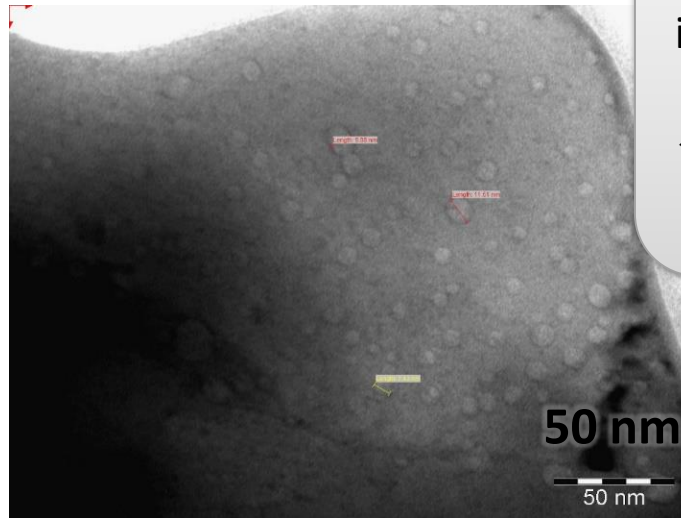
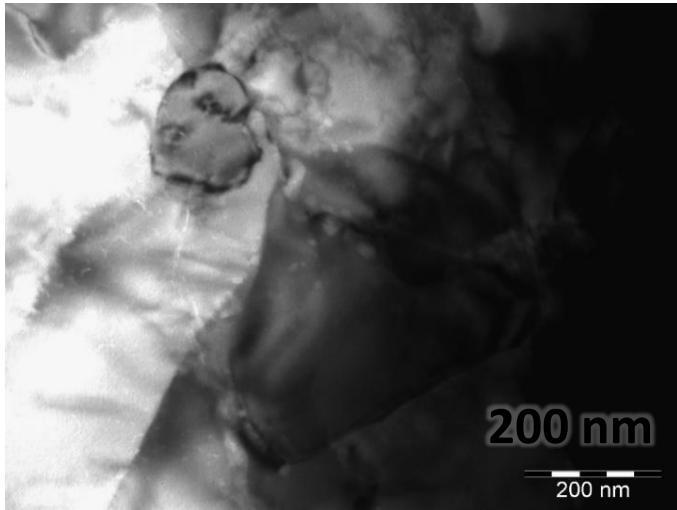
# TEM observation

'Bulk' of NiB layers – heat treated

Steel substrate



Directly on steel :  
dendritic growth



On NiP:  
isotropic grain growth  
with 2 populations:  
✓  $\text{Ni}_3\text{B}$  : 100- 200 nm  
✓ Ni or  $\text{Ni}_2\text{B}$ : 10 nm

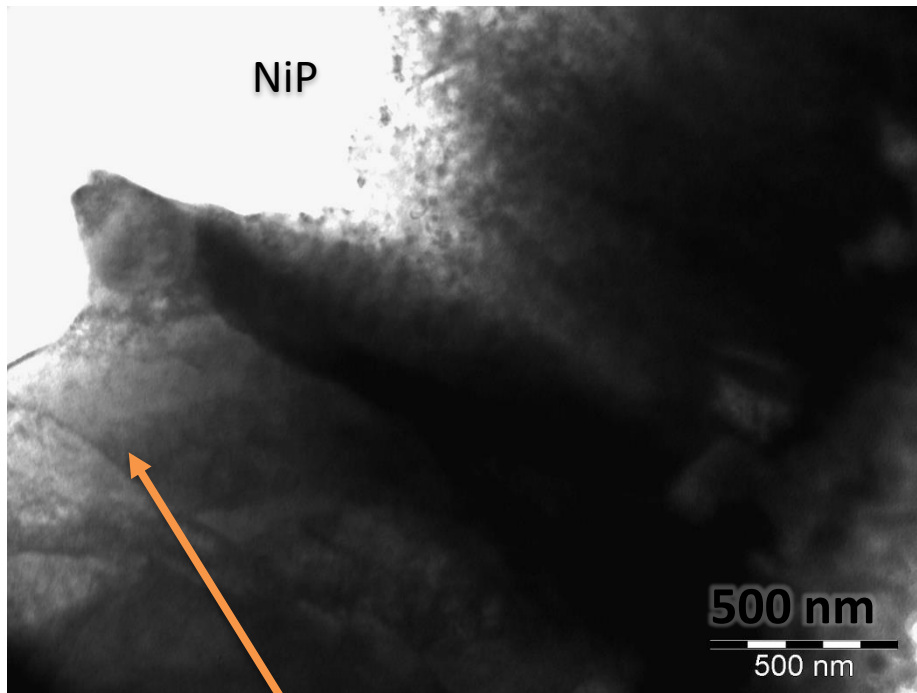
Linked with surface  
chemistry  
modification

# TEM observation

Interfaces

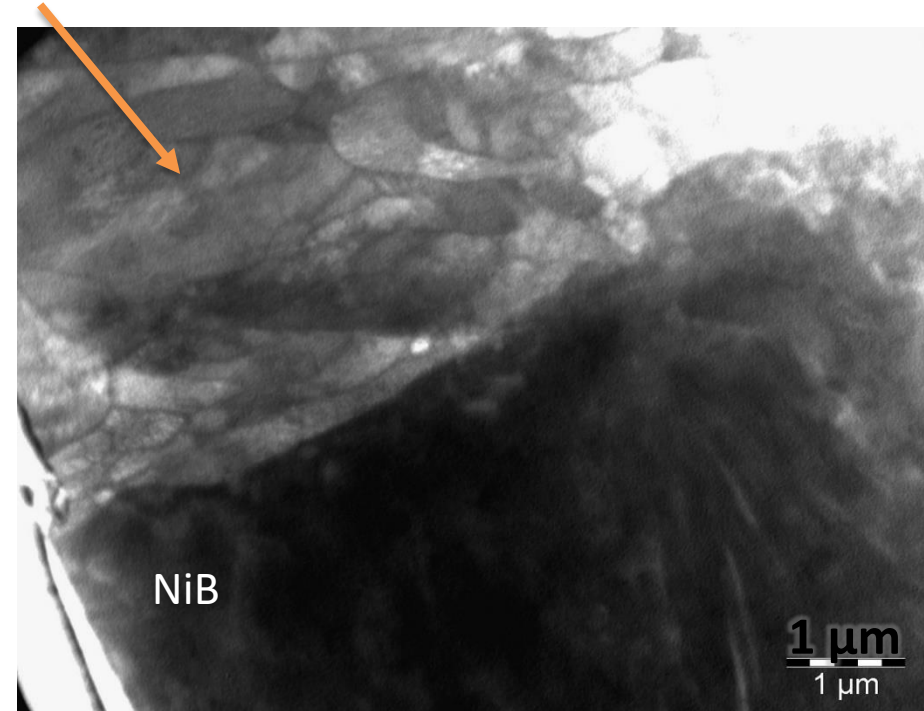
As plated

Steel substrate



Steel substrate

Isotropic growth

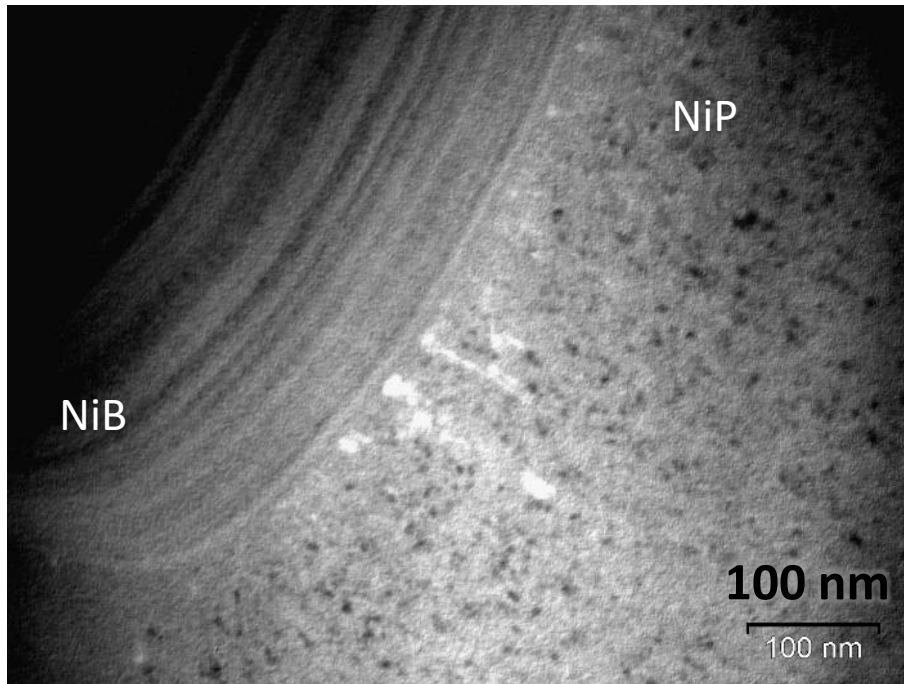


Feather-like  
morphology

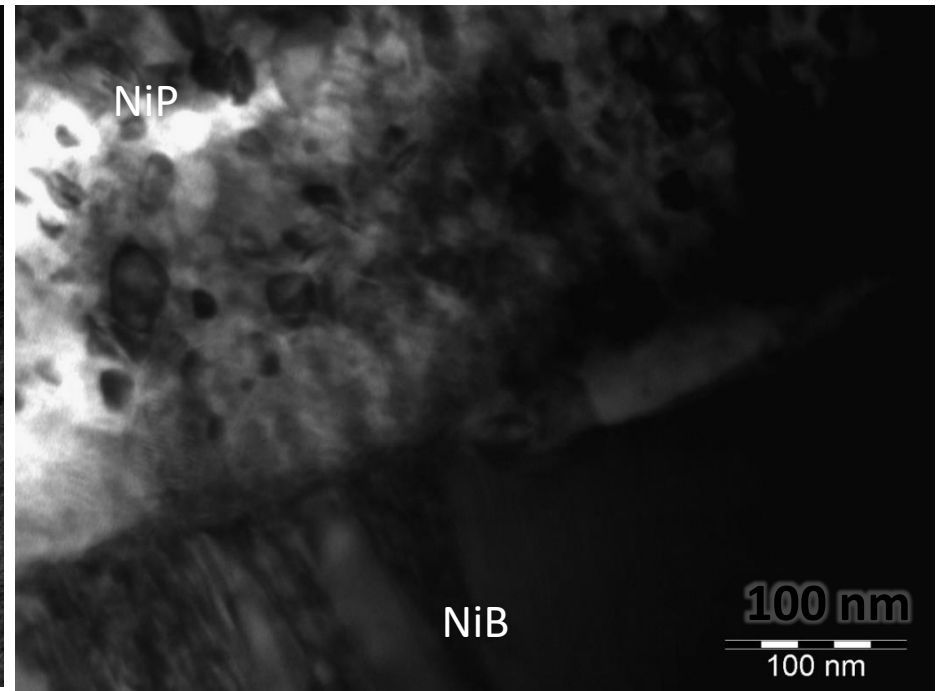
# TEM observation

## Interfaces

As plated



Heat treated

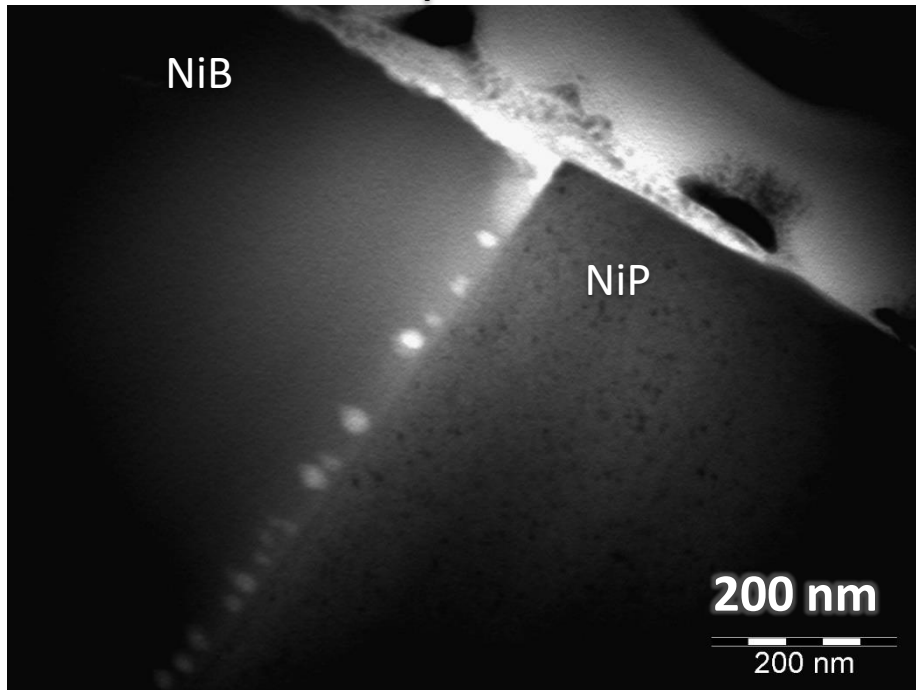


NiP on NiB:  
'wavy' interface  
Isotropic growth of NiP

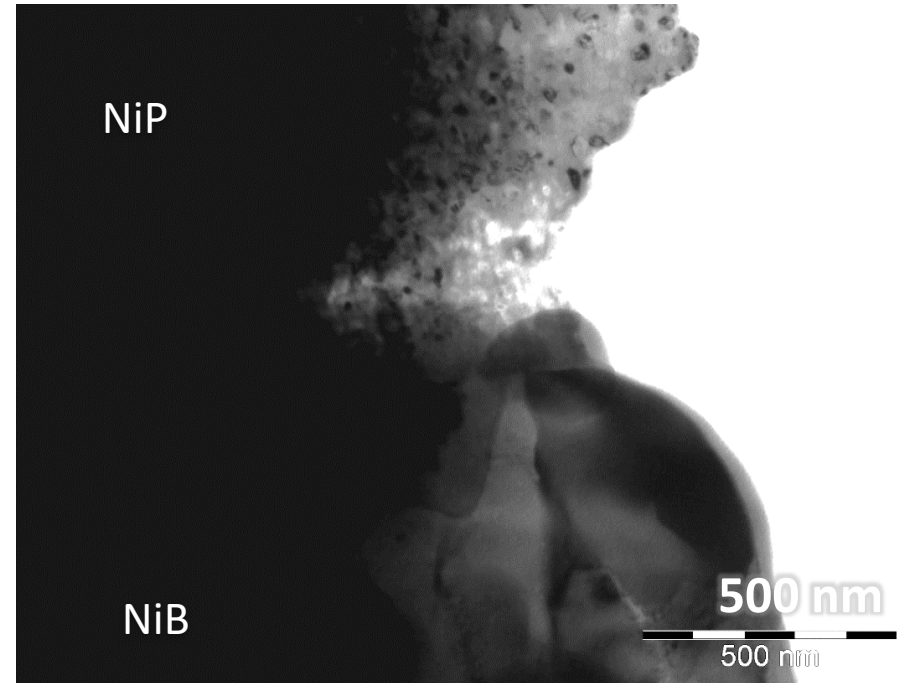
# TEM observation

## Interfaces

As plated



Heat treated



NiB on NiP:

‘flat’ interface

No observable features in as-deposited state – columnar growth after heat treatment



# Conclusions

- Modified FIB-SEM lift out method allows observation of interfaces
- Growth features from NiB present at small scale : waviness of profile and columnar features
- Influence on 'underlayer' on crystallization



**Acknowledgements: F. Tosar and S. Gobeaut**